





Komentarz o paśmie przewodnictwa

W zależności od półprzewodnika dno pasma przewodnictwa może być zbudowane z różnych dolin – ta sama hetereostruktura może być studnią w jednym paśmie (np. Γ) i barierą w innym (np. X)



AlAs, where X is the lowest minimum in the barrier. (c) Well of GaAs surrounded by AlAs.

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Potencjał harmonicz	zny
$\left[-\frac{\hbar^2}{2m}\frac{d^2}{dr^2} + \frac{1}{2}m\omega_0^2 z^2\right]\psi(z) = E\psi(z)$	$\varepsilon = \frac{E}{k}$ $\xi = \sqrt{\frac{m\omega_0}{k}}z$
$\left(\frac{d}{d\xi} - \xi\right) \left(\frac{d}{d\xi} + \xi\right) \operatorname{A} \operatorname{e}^{\frac{\xi^2}{2}} = (-2\varepsilon_0 + 1) \operatorname{A} \operatorname{e}^{\frac{\xi^2}{2}}$	$h\omega_0$ N h = 0 $\Rightarrow -2\varepsilon_0 + 1 = 0 \Rightarrow \varepsilon_0 = \frac{1}{2}$
	$\varepsilon_n = n + \frac{1}{2}$
$E_n = \hbar\omega_0 \left(n + \frac{1}{2} \right)$	
$\psi_n(z) = A_n H_n\left(\sqrt{\frac{m\omega_0}{\hbar}}z\right) \exp\left(-\frac{m\omega_0}{2\hbar}z^2\right)$	$\psi_3^2 \qquad \qquad$
H_n - wielomiany Hermite'a	$\psi_1^2 = \frac{E_1 = \frac{1}{2} \hbar \omega}{E_0 = \frac{1}{2} \hbar \omega}$
$A_n = \left(\sum^n n! \sqrt{m\omega} \right)$	-4 0 $\sqrt{\alpha x}$ 5
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Potencjał harmoniczny 2D
Szkic dowodu Funkcja Blocha nośnika w krysztale::
$\Psi(\vec{r}) = \sum_{n,k} c_{n,k} u_{n,k}(\vec{r}) e^{i\vec{k}\cdot\vec{r}}$
Dla elektronu:
$\Psi_{\rm c}(\vec{r}) \approx \sum_{k} c_{1,k} u_{\Gamma_6,0}(\vec{r}) e^{i\vec{k}\vec{r}} = u_{\Gamma_6,0}(\vec{r}) F_e(\vec{r})$
Dla dziury:
$\Psi_{\rm v}(\vec{r}) \approx \sum_{J_Z=\pm 3/2,\pm 1/2,k} c_{J_Z,k} u_{\Gamma_{\rm B},J_Z}(\vec{r}) e^{i\vec{k}\cdot\vec{r}} = \sum_{J_Z=\pm 3/2,\pm 1/2,k} u_{\Gamma_{\rm B},J_Z}(\vec{r}) F_{J_Z}(\vec{r})$
Dipolowe przejścia optyczne międzypasmowe:
$ \langle \Psi_{c}(\vec{r}) \vec{p} \Psi_{v,J_{z}}(\vec{r}) \rangle = \langle u_{\Gamma_{6},0}(\vec{r}) u_{\Gamma_{8},J_{z}}(\vec{r}) \rangle \langle F_{e}(\vec{r}) \vec{p} F_{J_{z}}(\vec{r}) \rangle + \langle u_{\Gamma_{6},0}(\vec{r}) \vec{p} u_{\Gamma_{8},J_{z}}(\vec{r}) \rangle \langle F_{e}(\vec{r}) F_{J_{z}}(\vec{r}) \rangle $
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госспсјати	annoniczny SD	
$E_n^x = \hbar\omega_0 \left(n_x + \frac{1}{2}\right) w$	kierunku <i>x, y</i> i <i>z</i>	
$E_n = E_n^x + E_n^y + E_n^z = \hbar$	$\omega_0\left(N+\frac{3}{2}\right)$	
Degeneracja? $g_N = \frac{(N+1)(N+2)}{2}$	$N = n_x + n_y + n_z$	
N	$(\boldsymbol{n}_{\boldsymbol{x}}, \boldsymbol{n}_{\boldsymbol{y}}, \boldsymbol{n}_{\boldsymbol{z}})$	
	(0,0,0)	
0		
0	(1,0,0) (0,1,0) (0,0,1)	
0 1 2	(1,0,0) (0,1,0) (0,0,1) (2,0,0) (0,2,0) (0,0,2) (1,1,0) (1,0,1) (0,1,1)	
0 1 2 3	(1,0,0) (0,1,0) (0,0,1) (2,0,0) (0,2,0) (0,0,2) (1,1,0) (1,0,1) (0,1,1) 3x(3,0,0) 1x(1,1,1) 6x(2,0,1)	

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Sferyczne kropki kwantowe

Przerwa energetyczna w sferycznych kropkach kwantowych [Brus, L. E. J. Phys. Chem. 1986, **90**, 2555, Brus. L. E. J. Chem. Phys. 1984, **80**, 4403]



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Sferyczne kropki kwantowe ematic illustration of bioconjugation (a) Bifunctional linkage (d) Electrostatic attraction nethods. (a) Use of a bifunctional ligand methods. (a) Use of a bifunctional ligand such as mercaptoacetic acid for linking QDs to biomolecules [8⁺⁺]. (b) TOPO-capped QDs bound to a modified acrylic acid polymer by hydrophobic forces. (c) QD solubilization and S-CH2-CO-NH-biomol bioconjugation using a mercaptosilane compound [7**]. (d) Positively charged compound (1⁻⁴), (0) Positively charged biomolecules are linked to negatively charged QDs by electrostatic attraction [9]. (e) Incorporation of QDs in microbeads and nanobeads [20**]. (b) Hydrophobic attraction (-cooh O=P_____HNOC COOH (e) Nanobeads (c) Silanization -CH2-CO-NH- biomolecu Current Opinion in Biotechnology Luminescent quantum dots for multiplexed biological detection and imaging W. Chan et al. Current Opinion in Biotechnology 2002, 13:40-46







DTERYCZ	IE Kropki kwantowe
	Photon energy (eV)
	4.0 3.0 2.0 1.0 0.8 0.7 0.6 0.5 0.4
	- 335 (het-0) - 335 (het-0) - 435 (het-0) - 438 (het-0) - 438 (het-0) - 438 (het-0) - 1054 (het-0) - 1054 (het-0) - 2540 (E::/MG) - 1054 (het-0) - 1054 (het
	CdS/ZnS CdSe CdTe CdTe/CdSe 20056 CdS CdSe CdTe/CdSe 20056 CdS CdSe CdTe/CdSe CdS CdS CdSe CdSe CdS CdS CdSe CdSe CdS CdS CdSe CdSe CdS CdS CdS CdS CdSe CdS CdS CdS CdS CdS CdS CdS CdS CdS CdS
	1 UV V IBI G 107 R Near-IR Mid-IR 300 400 500 600 700 800 1000 2000 3000 Emission wavelength (nm)
	Current Opinion in Chemical Biology









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