Lecture 1: Introduction Lecturer: Jeffrey Everts, 5.32 jeffrey.everts@fuw.edu.pl Tutor: Reinier van Buel, 5.63, r. van-buel @uw-edn.pl. Material: Lecture notes (in progress), books (see website). Website: www.fuw.edu.pl/~jeverts/teaching/statphysb Organisation: Lectures on Wednesday, 12:15-15:00, room 2.06. Tutorials on Thurstay, 16:15-19:00, room 1.38 Examination: Mid-term (20%), Final (60%), hand-ins (20%) What is Statistical Physics about! Microscopic interpretation: Typical physical system can be described by a Hamiltonian, e.g. H(pn; u)= = (p1, ..., pn)

H(pn; u)= = (p1, ..., pn)

For a classical system; we would like to know ? ph(t), rh(t) {

(phase space) determined by $\vec{r}_i = \frac{\partial H}{\partial \vec{r}_i}$; $\vec{p}_i = -\frac{\partial H}{\partial \vec{r}_i}$ Hamilton equations of motion.

In the case of guantum system: it $\frac{d}{dt} | \Psi(t) \rangle = \hat{H} | \Psi(t) \rangle$

Schrödinger equation (Hilbert space)

However, only limited number of problems are "solvable"

Single particle (⊕=0)

However, macroscopic

· Particle in external potential

systems N~ ro23...

· Two-particle problems with radial potential.
Glassical mechanics, gruantum mechanics.

Macroscopic interpretation:
Instead of 2ph(t), Th(t) [classical) or [X(t)) (quantum) the
macroscopic state is determined by just a few variables.
Kramples: N M p nN (chemical work, mass franspe p p TS (mechanical) (heart) F. E (clectric) extensive variables variables E E E E E E E E E E E E E
This is the field of (equilibrium) thermodynamics.
Roughly speaking how the available energy of a system is redistributed between heat and work. Classical Machanics, & Thermodynamics
Classical mechanics, guantum mechanics
microscopic macroscopic. statistical mechanics.
Pragnatic view on statistical mechanics: (Recall we assume elementare view on statistical mechanics: (Recall we assume elementare
Consider time and time $(\exists^{N}(t) \exists^{N}(t))$ or $ \Psi(t)\rangle$
System with some variables controlled. I independent measurements. Of observable O

$$O_{obs} = \frac{1}{N} \sum_{\alpha=1}^{N} O_{\alpha}$$

where we measure long enough (N+05) 2) system will flowarbitrarily close to all microscopic states consistent with constraints.

Then

$$O_{obs} = \sum_{v} \left[\frac{1}{N} \left(\text{Number of times state } v \text{ is } \right) O_{v} \right]$$

with $O_{v} = \langle v | O_{v} \rangle$.

with $O_{v} = \langle v | O_{v} \rangle$.

In a spent in state v .

$$\vec{O} = \int_{t_0}^{t_0+T} \int_{t_0}^{t_0+T} dt O(\Gamma(t))$$

For t sufficiently long 0=20> Ergodic system (not obvious!)

How do we construct Pr / f(r)? Fundamental assumption of statistical mechanics (cannot be proven)

for an isolated system with fixed E, V, N all microscopic states are equally likely at thermodynamic equilibrium.

Uniform distribution of microscopic states with same me rgy and system size. Equilibrium is the most "random" state.

For example for discrete energy levels P, = 1 (N,V,E) SE, EV.

Normalisation:
$$\sum_{v} P_{v} = 1$$
. (a) $\Omega(N_{1}V_{1}E) = \sum_{v} \delta_{E_{1}E_{v}}$.

Gondinuous (classical) case:
$$f_m(\Gamma) = \frac{\delta(E-H(\Gamma))}{\omega(E,V,N)}$$

Link to thermodynamics
$$S = k_B \ln \Omega(N_1 V_1 E)$$
.

La The collection of microstates subjected to macroscopic constraints is called an ensemble. E.g. (N,V,E) fixed #s called microcanonical

From the microcanonical ensemble other ensembles can be derived

For example, canonical ensemble. (N,V,T) fixed.

$$N_{\delta} + N_{R} = N$$

$$V_S + V_R = V$$
.

System is in equilibrium with bath: $\beta := k_B T = \left(\frac{\partial ln l}{\partial E}\right)_{NIV}$. We take reservoir in thermodynamic limit:

$$\frac{N_R}{N_S}$$
, $\frac{V_R}{V_S}$, $\frac{E_R}{E_S} \rightarrow \infty$
Suppose $E_S \approx E_D$.

=) ln
$$\Omega_R(E-E_y) = ln \Omega_R(E) - \left(\frac{\partial ln \Omega_R}{\partial E}\right)_{NR,VR} E_y +$$

Higher order derms vanish:

E.q.
$$\frac{\partial^2 \ln \Omega_R}{\partial E^2} = \sqrt{\frac{\partial \beta}{\partial E}}_{N,V} = -k_B \beta^2 \zeta_V^{-1} \rightarrow 0$$

extensive.

=> Py or exp (-BEy)

Z(N,V,T) = Zi e-BEV Ganonical partition function.

ZENIVIT) = Tr[e-BH].

2 trace over complete set of states.

Also note that:

$$\langle E \rangle = \sum_{i} P_{i} E_{i} = \sum_{j=1}^{N} \sum_{i=1}^{N} \frac{\partial}{\partial (-\beta_{i})} e^{-\beta E_{i}}$$

$$= \frac{1}{2} \frac{\partial}{\partial (-\beta_{i})} \sum_{i=1}^{N} e^{-\beta E_{i}} = \frac{1}{2} \frac{\partial}{\partial (-\beta_{i})} e^{-\beta E_{i}}$$
uniform convergent:
$$E = F + T \leq F$$

Compare: E=F+TS=F-T (2F) NIV=F+B(2F) NIV

=) BF = -ln 2 (N, V,T) hink to thermodynamics.

= (3/2) MIn.

Other ensembles:

Grand canonical ensemble (µ1 VIT) fixed.

\[\begin{align*}
& \begin{align

with link to thermodynamics: BS (MVIT) = -ln \(\ext{(u, ViT)}\),

 $\Omega(\mu, V, T) = P(\langle N \rangle, V, T) - \mu \langle N \rangle$ (grand potential)

So we see always that:

complete thermodynamics via e.g. dD=-SdT-pdV-Ndy.

So we could say, the main problem of stat mech is to compute the partition function.

In elementary course of stat mech, you have seen how to compute partition functions for simple cases.

- · I deal systems (classical and quantum)
- · Some simple models, like (ID) Lattice gases, Ising model.

In Stat Phys B we take it a step further:

* Partition functions of interacting systems

mostly classical because for quantum systems
more tools are needed, e.g. second quantisation.

- * Phase transitions.
- & Some elementary concepts in non-equilibrium systems.

Placement

- · Quantum mang-body sheary.
- · Critical Phenomera.
- · Non-eg. Stat phys.