

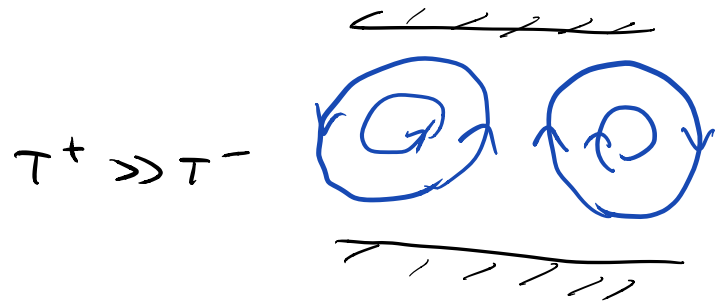
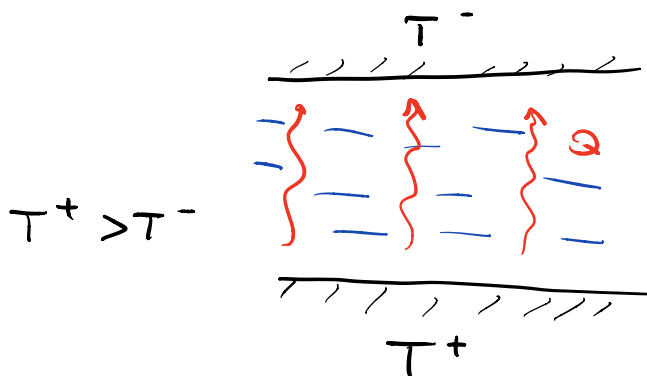
CONVECTION

- 1) Heat - major driving factor
 - hurricanes, cyclones, tornadoes
 - continental drift
 - stellar transport
 - central heating
- 2) Convection is a combination of gravity & material properties - most fluids expand when heated (density decreases with T)
- 3) Buoyancy forces set the fluid in motion.

Examples:

(A) steady laminar convection flow

(B) Thermal instability - Rayleigh-Bénard convection



Note Concentration gradients in mixed fluids can also cause convective flows (sometimes called advective flows) (ocean)

We need to incorporate new variables into the eqs. of motion.

Heat & motion are coupled PDE.

(Typically v. hard to solve analytically)

→ approximations needed.

Thermal expansion coefficient

isobaric th. exp. coeff.

$$\alpha = -\frac{1}{\rho} \left(\frac{\partial \rho}{\partial T} \right)_p$$

- universal 'constant'

- for a gas

$$pV = nRT$$

$$\Rightarrow \rho \sim \frac{1}{V} \propto \frac{P}{T} \quad \rho \sim \frac{1}{T}$$

$$\Rightarrow \alpha = \frac{1}{T}$$

$$\text{air} \quad \alpha \approx 3 \cdot 10^{-3} \frac{1}{K} \quad @ \text{ room } T.$$

- for liquids similarly

$$\text{water} \quad \alpha \approx 2.5 \cdot 10^{-4} \frac{1}{K} \quad @ 25^\circ C$$

The resulting change in density

$$\Delta \rho = -\alpha \rho \Delta T$$

for $|\alpha \Delta T| \ll 1$

→ extra buoyancy term $\Delta \rho g$ in N-S eqs.

For a flow with U , L , temperature variation Θ ,
the ratio of buoyancy to advective terms

$$Ri = \frac{|\Delta \rho g|}{|\rho(\mathbf{v} \cdot \nabla) \mathbf{v}|} = \frac{\alpha \rho \Theta g}{\rho \frac{U^2}{L}} = \frac{\alpha \Theta g}{U^2} L$$

Richardson number

Ri small - advection dominates

$Ri \sim 1 \Rightarrow$ flow driven by convection with typical speed $U \sim \sqrt{\alpha \Theta} \sqrt{gL}$

\uparrow small \uparrow free-fall velocity from $\frac{L}{2}$.

HEAT IN FLUIDS

Heat is transported in a fluid \rightarrow laws of thermodynamics

First law

$$\Delta \mathcal{E} = \underset{\text{heat}}{Q} + \underset{\text{work}}{W} \quad (\text{interaction with environment})$$

Evolution of energy of a fluid particle.

$$\frac{D\mathcal{E}}{Dt} = \underset{\substack{\uparrow \\ \text{heat transfer} \\ \text{rate from the} \\ \text{environment}}}{\dot{Q}} + \underset{\substack{\uparrow \\ \text{rate of work performed} \\ \text{on the fluid element.}}}{\dot{W}}$$

$$\mathcal{E} = \int_V \rho \mathcal{E} dV \quad \rho \mathcal{E} = \frac{d\mathcal{E}}{dV} \quad \text{energy density}$$

Reynolds theorem

$$\frac{D\varepsilon}{Dt} = \frac{d\varepsilon}{dt} + \oint_{S=\partial V} p \varepsilon (\underline{v} - \underline{v}_s) \cdot d\underline{S}$$



↑
energy
production

↑ net outflow from the control volume.

Rate of work

(see Landrup for details)

through contact forces on the surface of control volume

$$\dot{W} = \oint_S \underline{\sigma} \cdot \underline{\underline{v}} \cdot d\underline{S}$$

Rate of heat transfer

- conduction (diffusion)
- advection (convection)
- radiation

Current density $\underline{q}(\underline{x}, t)$ such that the amount of heat through $d\underline{S}$ in a time δt is $\delta Q = \underline{q} \delta t \cdot d\underline{S}$

The total rate of heat transfer into the system because

$$\dot{Q} = - \oint_S \underline{q} \cdot d\underline{S} + \int_V h dV$$

(*)

↑ internal heat production rate
(nuclear, chemical or other)

Internal energy and heat equation

$$\varepsilon = \tau + \mathcal{V} + u$$

↑ kinetic

$$\tau = \int_V \frac{1}{2} \rho \underline{v}^2 dV$$

↑ potential

$$\mathcal{V} = \int_V \rho \Phi dV$$

↑ internal energy

$$u = \int_V \rho u dV$$

$$\epsilon = \frac{1}{2} v^2 + \Phi + u$$

$$\frac{D\epsilon}{Dt} = \frac{D\epsilon}{Dt} - \frac{D\tau}{Dt} - \frac{D\psi}{Dt} = \dot{W} + \dot{Q} - \dot{P}$$

total power
without gravity
contribution

$$\frac{D\epsilon}{Dt} = \dot{Q} + \dot{W}_{int} \quad \left(\begin{array}{l} \text{work against} \\ \text{internal stresses} \\ \text{(dissipation)} \end{array} \right)$$

(*) transferred into a volume integral:

$$\int_V \frac{D\epsilon}{Dt} dV = \int_V \left(- \underbrace{\nabla \cdot \underline{q}}_{\text{heat conduction}} + \underbrace{h}_{\text{heat production}} + \underbrace{\sum_{ij} \sigma_{ij} \nabla_j v_i}_{\sum_{ij} \epsilon_{ij} \epsilon_{ij}} \right) dV$$

HEAT EQUATION

rate of dissipation
(density)

Heat flux

Second Law implies that heat flows from high T to low T regions.

The law by which this happens is empirical.

For an isotropic medium - Fourier's law of heat conduction

$$\underline{q} = -k \nabla T$$

↑
thermal conductivity

water $k \sim 0.6 \text{ W/Km}$

air $k \sim 0.025 \text{ W/Km}$

Fourier's equation

thermodynamic variable $T(\underline{r}, t)$

specific energy $u = u(T)$ is a local fun of T .

We assume $u = c_0 T$
specific heat capacity

for a fluid at rest ($\underline{v} = 0$) the heat eq. becomes

$$\rho c_0 \frac{\partial T}{\partial t} = k \nabla^2 T + h$$

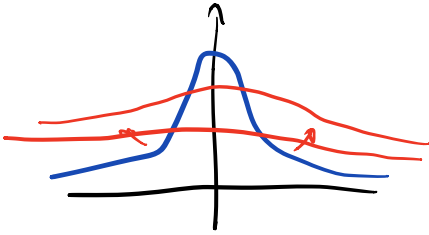
conduction source

if $h=0$

$$\frac{\partial T}{\partial t} = \kappa \nabla^2 T$$

Fourier's diffusion equation

$$\kappa = \frac{k}{\rho c_0} \quad \text{heat diffusivity}$$



water

$$\kappa = 1.4 \cdot 10^{-7} \frac{\text{m}^2}{\text{s}}$$

$$\kappa \sim \frac{\nu}{6} \quad \left(\frac{1}{6} \text{ of momentum diffusivity} \right)$$

For a moving fluid we have (approximately)

$$\frac{DT}{Dt} = \frac{\partial T}{\partial t} + (\underline{v} \cdot \nabla) T = k \nabla^2 T$$

advection diffusion

advection time

$$t_{adv} \sim \frac{L}{u}$$

diffusion time

$$t_{diff} \sim \frac{L^2}{\kappa}$$

$$\left\{ \begin{array}{l} Pe \rightarrow \infty \text{ advection-dominated} \\ Pe \rightarrow 0 \text{ diffusion-dominated} \end{array} \right.$$

Péclet number

$$Pe \sim \frac{t_{adv}}{t_{diff}}$$

$$Pe = \frac{|\underline{v} \cdot \nabla T|}{|\kappa \nabla^2 T|} \approx \frac{uL}{\kappa}$$

Note Full heat eq.

$$\rho c_p \left(\frac{\partial T}{\partial t} + (\underline{v} \cdot \nabla) T \right) = \underbrace{\kappa \nabla^2 T}_{\text{temp. diffusion}} + \underbrace{h}_{\text{heat production}} + \underbrace{2\eta \underline{e} : \underline{e} + \gamma (\nabla \cdot \underline{v})^2}_{\text{viscous dissipation}}$$

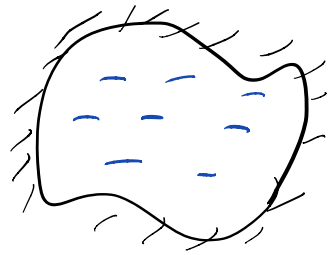
shear viscosity
↓
bulk viscosity
↓
(only for compressible)

$\frac{\eta}{\rho c_p} \underline{e} : \underline{e}$ is typically small.

The Boussinesq approximation

$$\underline{g} = (0, 0, -g)$$

- incompressible fluid ρ_0
- temperature T_0
- hydrostatic pressure $p = p_0 - \rho_0 g z$



At a certain time boundary temperatures are changed and a flow \underline{v} is created.

Main assumption: thermal variations are small

i.e. $|\alpha \Delta T| \ll 1$

$$\Delta T = T - T_0$$

⇒ we can write

$$\Delta p = - \rho_0 \alpha \Delta T$$

↑
const.

We add to N-S - term

$$\Delta p \underline{g} = - \rho_0 \alpha \Delta T \underline{g}$$

We modify

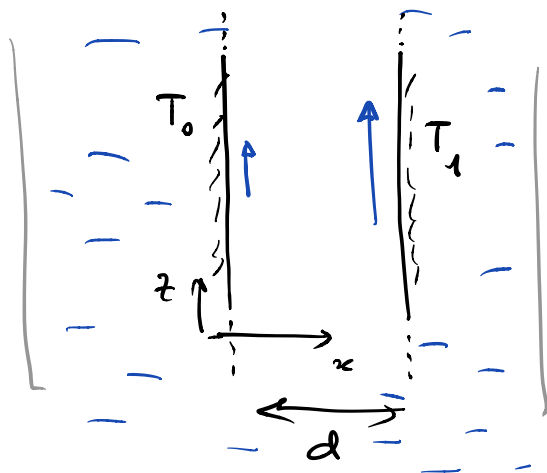
$$p = p_0 - \rho_0 g z + \Delta p$$

to get:

Our final eqs:

$$\begin{cases} \frac{\partial \Delta T}{\partial t} + (\underline{v} \cdot \nabla) \Delta T = \kappa \nabla^2 \Delta T \\ \frac{\partial \underline{v}}{\partial t} + (\underline{v} \cdot \nabla) \underline{v} = - \frac{\nabla \Delta p}{\rho_0} + \nu \nabla^2 \underline{v} - \alpha \Delta T \underline{g} \\ \nabla \cdot \underline{v} = 0 \end{cases}$$

{ Steady convection in an open vertical slot heated on one side



$$T_1 > T_0$$

Steady flow of heat & mass

$$\underline{v} = (0, 0, v_z(z))$$

$$T = T(x)$$

(long plates)

all in a reservoir of fluid

Equations of motion

Heat eq. $\frac{\partial}{\partial t} = 0$ ($(\underline{v} \cdot \nabla) = 0$ unidirectional flow)

$$\nabla^2 T = 0 \quad (\Rightarrow) \quad \frac{d^2}{dx^2} T(x) = 0$$

$$T = T_0 + \Delta T = T_0 + \Theta \frac{x}{d} \quad \Theta = T_1 - T_0$$

static profile

N-S $\nabla_x \Delta p = \nabla_z \Delta p = 0$

$$\Delta p = p(z)$$

The z -component:

$$\underbrace{\frac{1}{\rho_0} \frac{\partial}{\partial z} \Delta p(z)}_{\text{dep. on } z} = \underbrace{\nu \frac{\partial^2}{\partial x^2} v_z(x)}_{\text{dep. on } x} + \alpha \Theta \frac{x}{d} g$$

both have to be const.

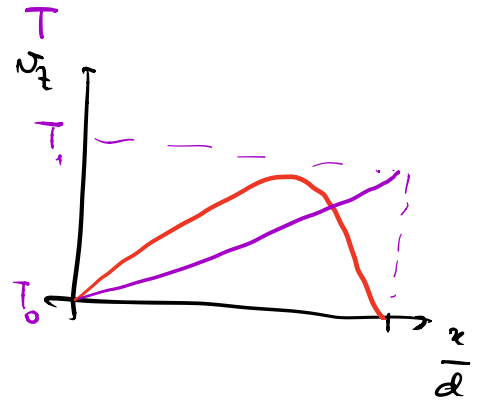
$$1) \Delta p = 0$$

$$2) \nu v_z''(x) + \alpha \Theta \frac{x}{d} g = 0$$

+ b.c.

$$v_z(0) = 0$$

$$v_z(d) = 0$$



we find

$$v_z = \frac{\alpha \Theta g d^2}{6\nu} \frac{x}{d} \left(1 - \frac{x^2}{d^2} \right)$$

$$\cdot \text{max } v \text{ at } x = \frac{d}{\sqrt{3}}$$

$$\text{Avg. velocity} \quad U = \frac{1}{d} \int_0^d v_z(x) dx = \frac{\alpha \Theta g d^2}{24\nu}$$

Reynolds number

$$Re = \frac{Ud}{\nu} = \frac{1}{24} \frac{\alpha \Theta d^3 g}{\nu^2}$$

$$\underline{\text{Ex.}} \cdot \text{water} \quad d = 1 \text{ cm} \quad \Theta = 10 \text{ K} \rightarrow U \sim 12 \frac{\text{cm}}{\text{s}}$$

$$Re = 1400 \quad (\text{laminar})$$

$$\cdot \text{air} \quad U \sim 8 \text{ cm/s} \quad Re \approx 50.$$

Entrance length for heat

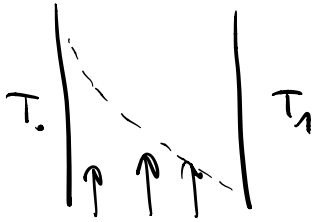
The rate at which heat is transported to the reservoir is the entire internal energy of the fluid at exit:

$$\dot{Q} = \int_0^d dx (pc \Delta T) v_z = \frac{\rho c \alpha \Theta^2 g d^3 L}{45\nu}$$

Mass conservation forces the fluid to enter at avg. velocity U . Heat takes $t \sim \frac{d^2}{4\kappa}$ to diffuse across the slot.

By that time the fluid will have moved by

$$l = Ut = \frac{\alpha \Theta g d^4}{96 \kappa \nu} = \frac{Ra}{96} d$$



$$Ra = \frac{\alpha \Theta g d^3}{\nu \kappa}$$

Rayleigh number

thermal transport $\frac{\text{diffusion}}{\text{convection}}$

Ex. $Ra = 2 \cdot 10^5$ $l = 21 \text{ m}$ (water)

$Ra = 900$ $l = 10 \text{ cm}$ (air)